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consolidate concepts learnt in class.

Students are also provided with

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Answers

Exploring Equilibrium Lab Pre-Lab

Questions 1. False, The reactions continue, but there is an equal balance of opposing reaction rates 2. False, the concentration of reactants and products remain constant with time. They don ' t necessarily have to

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be equal. 3. (a) When the humidity is low, the paper will be blue.

Color lab - Exploring Equilibrium Lab
Pre-Lab Questions 1 ...

Founded in 2002 by Nobel Laureate
Carl Wieman, the PhET Interactive
Simulations project at the University

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of Colorado Boulder creates free interactive math and science simulations. PhET sims are based on extensive education research and engage students through an intuitive, game-like environment where students learn through exploration and

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discovery.

Exploring Equilibrium - PhET
Contribution
via YouTube Capture

Exploring Equilibrium Mini Lab Part A
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The amount of reactants decreases and the amount of products increases in order to reach equilibrium. b) Describe the nature of dynamic equilibrium when small numbers of particles (such as 50, as compared to 6.022×10^{23}) are present. When small amounts of particles are present,

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equilibrium is not reached immediately; rather, it is a slow process. However, the reaction progresses at a constant, steady rate.

Equilibrium Lab | Chemical
Equilibrium | Reaction Rate

1. When red-colored $\text{Co}(\text{NO}_3)_2 \times 6\text{H}_2\text{O}$

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crystals are dissolved in 3mL of deionized water, the solution turns pink. To explain this, the equilibrium stress is on the product ' s side (addition of water), so the solution shifts towards to reactants. Since the reactants are favored, the 2.

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Lab 5- Chemical Equilibrium and Le Chatelier ' s Principle ...

In this lab, the effect of applying stresses to a variety of chemical systems at equilibrium will be explored. The equilibrium systems to be studied are given below: Saturated Sodium Chloride Solution \uparrow /ce{NaCl}

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(s) \rightarrow $\text{Na}^{+1}(\text{aq}) + \text{Cl}^{-1}(\text{aq})$ /
Acidified Chromate Solution

12: Equilibrium and Le Chatelier's
Principle (Experiment ...
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Exploring Equilibrium: It Works Both
Ways Student Laboratory Kit The
word equilibrium has two roots: aqul,
meaning equal, and libra, meaning
weight or hulance. Our physical sense
of equ librium suggests a condition of

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equal balance of opposing forces
How does this physical sense of
equilibrium translate to chemical
equilibrium? The purpose of this lab is
to explore the nature and
consequences of equilibrium in a
variety of chemical reactions
Chemical
Concepts .

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Solved: Exploring Equilibrium: It
Works Both Ways Student ...

Describe the relative sizes of the
forward and reverse rates at
equilibrium. Explain what effects
whether the equilibrium position
favors the products or the reactants.

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Predict how addition of a reactant or product will affect the forward and reverse reaction rates, and once this new system reaches equilibrium how the reactant and product concentrations will compare to the original system at ...

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Reactions & Rates - Reaction |
Kinematics | Concentration ...

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Equilibrium It Works Both Ways

Solution $\text{NaCl(s)} \rightleftharpoons \text{Na}^+(\text{aq}) + \text{Cl}^-$
1(aq) 12: Equilibrium and Le

Chatelier's Principle (Experiment...

Equilibrium is a condition in which all

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the forces in a system counteract each other. When adding vectors, equilibrium can be showed by a representative force, called an equilibrant. The

Lab Report On Exploring Equilibrium
It Works Both Ways

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Here you will test the effects of changing the volume and temperature on the complex ion equilibrium between $\text{Co}(\text{H}_2\text{O})_6^{2+}(\text{aq})$ and $\text{CoCl}_4^{2-}(\text{aq})$ as in Reaction 3.14. Reagents needed for this part are: $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}(\text{s})$, 12 M $\text{HCl}(\text{aq})$ (do not remove from fume hood), and

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deionized water. Reheat your hot-water bath from part B to a near boil.

3: Le Chatelier's Principle
(Experiment) - Chemistry ...
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