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arranged within the periodic table into groups and periods. CH 5 LAB: PERIODIC LAW (CHEM.A.2.3) Purpose

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arrange the W elements from a scrambled periodic table. You will also predict values for any information missing from the table. Theory: The present organization of the elements is a product of the Page 21/47

first periodic table published by Dmitri Mendeleev in 1869.

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Mendeleev Page 31/47

created the W first periodic table and was shortly followed by Meyer. They both arranged the elements by their mass and proposed that certain properties periodically reoccur. Meyer formed his Page 32/47

periodic law based on the atomic volume or molar volume, which is the atomic mass divided by the density in solid form.

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electron aw configuration. 4. Compare period and group trends of several properties. 5. Relate period and group trends in atomic radii to electron configuration. Discovery Lab: Versatile Metals, p. 151 Page 39/47

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several elements based on the elements' locations in the table.

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